# **Resene Paints (Australia) Limited**

Version No: 3.7

Safety Data Sheet according to WHS Regulations (Hazardous Chemicals) Amendment 2020 and ADG requirements

Chemwatch Hazard Alert Code: 4

Issue Date: **10/11/2022** Print Date: **10/11/2022** S.GHS.AUS.EN

### SECTION 1 Identification of the substance / mixture and of the company / undertaking

#### **Product Identifier**

Product name	Altex Thinning Solvent #213	
Synonyms	ns Not Available	
Proper shipping name PAINT (including paint, lacquer, enamel, stain, shellac, varnish, polish, liquid filler and liquid lacquer base) or PAINT RELATED MATERIAL (including paint thinning or reducing compound)		
Other means of identification	Not Available	

## Relevant identified uses of the substance or mixture and uses advised against

Relevant identified uses Thinning Solvent for Industrial Coatings

## Details of the manufacturer or supplier of the safety data sheet

Registered company name	Resene Paints (Australia) Limited	Altex Coatings Ltd
Address	7 Production Avenue, Molendinar Queensland 4214 Australia	91-111 Oropi Road Tauranga 3112 New Zealand
Telephone	+61 7 55126600	+64 7 541 1221
Fax	+61 7 55126697	+64 7 541 1310
Website	www.resene.com.au	www.altexcoatings.com
Email	Not Available	neil.debenham@carboline.co.nz

#### Emergency telephone number

Association / Organisation	AUSTRALIAN POISONS CENTRE	NZ POISONS (24hr 7 days)	CHEMWATCH EMERGENCY RESPONSE
Emergency telephone numbers	131126	0800 764766	+61 1800 951 288
Other emergency telephone numbers	Not Available	Not Available	+61 3 9573 3188

Once connected and if the message is not in your preferred language then please dial 01

### **SECTION 2 Hazards identification**

#### Classification of the substance or mixture

HAZARDOUS CHEMICAL. DANGEROUS GOODS. According to the WHS Regulations and the ADG Code.		
Poisons Schedule Not Applicable		
Classification <sup>[1]</sup>	Serious Eye Damage/Eye Irritation Category 2A, Specific Target Organ Toxicity - Single Exposure (Narcotic Effects) Category 3, Hazardous to the Aquatic Environment Long-Term Hazard Category 2, Reproductive Toxicity Category 2, Specific Target Organ Toxicity - Repeated Exposure Category 2, Flammable Liquids Category 2, Acute Toxicity (Inhalation) Category 4, Acute Toxicity (Oral) Category 4, Skin Corrosion/Irritation Category 2, Sensitisation (Skin) Category 1	
Legend:	1. Classified by Chemwatch; 2. Classification drawn from HCIS; 3. Classification drawn from Regulation (EU) No 1272/2008 - Annex VI	

Label elements

Hazard pictogram(s)	
Signal word	Danger

Hazard statement(s)

H319	Causes serious eye irritation.	
H336	May cause drowsiness or dizziness.	
H411	Toxic to aquatic life with long lasting effects.	
H361d	Suspected of damaging the unborn child.	
H373	May cause damage to organs through prolonged or repeated exposure.	

H225	Highly flammable liquid and vapour.
H332	Harmful if inhaled.
H302	Harmful if swallowed.
H315	Causes skin irritation.
H317	May cause an allergic skin reaction.

### Supplementary statement(s)

Not Applicable

## Precautionary statement(s) Prevention

P210	Keep away from heat, hot surfaces, sparks, open flames and other ignition sources. No smoking.
P233	Keep container tightly closed.
P260	Do not breathe mist/vapours/spray.
P271	Use only a well-ventilated area.
P280	Wear protective gloves, protective clothing, eye protection and face protection.
P240	Ground and bond container and receiving equipment.
P241	Use explosion-proof electrical/ventilating/lighting/intrinsically safe equipment.
P242	Use non-sparking tools.
P243	Take action to prevent static discharges.
P264	Wash all exposed external body areas thoroughly after handling.
P270	Do not eat, drink or smoke when using this product.
P273	Avoid release to the environment.
P272	Contaminated work clothing should not be allowed out of the workplace.

### Precautionary statement(s) Response

P308+P313	IF exposed or concerned: Get medical advice/ attention.
P370+P378	In case of fire: Use alcohol resistant foam or normal protein foam to extinguish.
P302+P352	IF ON SKIN: Wash with plenty of water and soap.
P305+P351+P338	IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing.
P333+P313	If skin irritation or rash occurs: Get medical advice/attention.
P337+P313	If eye irritation persists: Get medical advice/attention.
P362+P364	Take off contaminated clothing and wash it before reuse.
P391	Collect spillage.
P301+P312	IF SWALLOWED: Call a POISON CENTER/doctor/physician/first aider if you feel unwell.
P303+P361+P353	IF ON SKIN (or hair): Take off immediately all contaminated clothing. Rinse skin with water [or shower].
P304+P340	IF INHALED: Remove person to fresh air and keep comfortable for breathing.
P330	Rinse mouth.

## Precautionary statement(s) Storage

P403+P235	Store in a well-ventilated place. Keep cool.
P405	Store locked up.

### Precautionary statement(s) Disposal

P501 Dispose of contents/container to authorised hazardous or special waste collection point in accordance with any local regulation.

### **SECTION 3 Composition / information on ingredients**

#### Substances

See section below for composition of Mixtures

### Mixtures

CAS No	%[weight]	Name
108-88-3	80-90	toluene
1330-20-7	1-10	xylene
Legend:	1. Classified by Chemwatch; 2. Classification drawn from HCIS; 3. Classification drawn from Regulation (EU) No 1272/2008 - Annex VI; 4. Classification drawn from C&L * EU IOELVs available	

# **SECTION 4 First aid measures**

Description of first aid measures		
Eye Contact	If this product comes in contact with the eyes: <ul> <li>Wash out immediately with fresh running water.</li> <li>Ensure complete irrigation of the eye by keeping eyelids apart and away from eye and moving the eyelids by occasionally lifting the upper</li> </ul>	

	<ul> <li>and lower lids.</li> <li>Seek medical attention without delay; if pain persists or recurs seek medical attention.</li> <li>Removal of contact lenses after an eye injury should only be undertaken by skilled personnel.</li> </ul>
Skin Contact	If skin contact occurs: <ul> <li>Immediately remove all contaminated clothing, including footwear.</li> <li>Flush skin and hair with running water (and soap if available).</li> <li>Seek medical attention in event of irritation.</li> </ul>
Inhalation	<ul> <li>If fumes or combustion products are inhaled remove from contaminated area.</li> <li>Lay patient down. Keep warm and rested.</li> <li>Prostheses such as false teeth, which may block airway, should be removed, where possible, prior to initiating first aid procedures.</li> <li>Apply artificial respiration if not breathing, preferably with a demand valve resuscitator, bag-valve mask device, or pocket mask as trained. Perform CPR if necessary.</li> <li>Transport to hospital, or doctor, without delay.</li> </ul>
Ingestion	<ul> <li>If spontaneous vomiting appears imminent or occurs, hold patient's head down, lower than their hips to help avoid possible aspiration of vomitus.</li> <li>If swallowed do NOT induce vomiting.</li> <li>If vomiting occurs, lean patient forward or place on left side (head-down position, if possible) to maintain open airway and prevent aspiration.</li> <li>Observe the patient carefully.</li> <li>Never give liquid to a person showing signs of being sleepy or with reduced awareness; i.e. becoming unconscious.</li> <li>Give water to rinse out mouth, then provide liquid slowly and as much as casualty can comfortably drink.</li> <li>Seek medical advice.</li> <li>Avoid giving milk or oils.</li> <li>Avoid giving alcohol.</li> </ul>

#### Indication of any immediate medical attention and special treatment needed

Any material aspirated during vomiting may produce lung injury. Therefore emesis should not be induced mechanically or pharmacologically. Mechanical means should be used if it is considered necessary to evacuate the stomach contents; these include gastric lavage after endotracheal intubation. If spontaneous vomiting has occurred after ingestion, the patient should be monitored for difficult breathing, as adverse effects of aspiration into the lungs may be delayed up to 48 hours.

### **SECTION 5 Firefighting measures**

#### Extinguishing media

- Foam.
- Dry chemical powder.
- BCF (where regulations permit).Carbon dioxide.
- Water spray or fog Large fires only.

### Special hazards arising from the substrate or mixture

Fire Incompatibility	Avoid contamination with oxidising agents i.e. nitrates, oxidising acids, chlorine bleaches, pool chlorine etc. as ignition may result
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#### Advice for firefighters

Fire Fighting	<ul> <li>Alert Fire Brigade and tell them location and nature of hazard.</li> <li>May be violently or explosively reactive.</li> <li>Wear breathing apparatus plus protective gloves in the event of a fire.</li> <li>Prevent, by any means available, spillage from entering drains or water course.</li> <li>Consider evacuation (or protect in place).</li> <li>Fight fire from a safe distance, with adequate cover.</li> <li>If safe, switch off electrical equipment until vapour fire hazard removed.</li> <li>Use water delivered as a fine spray to control the fire and cool adjacent area.</li> <li>Avoid spraying water onto liquid pools.</li> <li>Do not approach containers suspected to be hot.</li> <li>Cool fire exposed containers with water spray from a protected location.</li> <li>If safe to do so, remove containers from path of fire.</li> </ul>
Fire/Explosion Hazard	<ul> <li>Liquid and vapour are highly flammable.</li> <li>Severe fire hazard when exposed to heat, flame and/or oxidisers.</li> <li>Vapour may travel a considerable distance to source of ignition.</li> <li>Heating may cause expansion or decomposition leading to violent rupture of containers.</li> <li>On combustion, may emit toxic fumes of carbon monoxide (CO).</li> <li>Combustion products include:</li> <li>carbon dioxide (CO2)</li> <li>other pyrolysis products typical of burning organic material.</li> </ul>
HAZCHEM	•3YE

#### **SECTION 6 Accidental release measures**

Personal precautions, protective equipment and emergency procedures

# See section 8

**Environmental precautions** 

See section 12

Minor Spills	<ul> <li>Remove all ignition sources.</li> <li>Clean up all spills immediately.</li> <li>Avoid breathing vapours and contact with skin and eyes.</li> <li>Control personal contact with the substance, by using protective equipment.</li> <li>Contain and absorb small quantities with vermiculite or other absorbent material.</li> <li>Wipe up.</li> <li>Collect residues in a flammable waste container.</li> </ul>
Major Spills	<ul> <li>Clear area of personnel and move upwind.</li> <li>Alert Fire Brigade and tell them location and nature of hazard.</li> <li>May be violently or explosively reactive.</li> <li>Wear breathing apparatus plus protective gloves.</li> <li>Prevent, by any means available, spillage from entering drains or water course.</li> <li>Consider evacuation (or protect in place).</li> <li>No smoking, naked lights or ignition sources.</li> <li>Increase ventilation.</li> <li>Stop leak if safe to do so.</li> <li>Water spray or fog may be used to disperse /absorb vapour.</li> <li>Contain spill with sand, earth or vermiculite.</li> <li>Use only spark-free shovels and explosion proof equipment.</li> <li>Collect recoverable product into labelled drums for disposal.</li> <li>Wash area and prevent runoff into drains.</li> <li>If contamination of drains or waterways occurs, advise emergency services.</li> </ul>

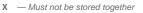
Personal Protective Equipment advice is contained in Section 8 of the SDS.

### **SECTION 7 Handling and storage**

Precautions for safe handling	
Safe handling	<ul> <li>Containers, even those that have been emptied, may contain explosive vapours.</li> <li>Do NOT cut, drill, grind, weld or perform similar operations on or near containers.</li> <li>Avoid all personal contact, including inhalation.</li> <li>Wear protective clothing when risk of exposure occurs.</li> <li>Use in a well-ventilated area.</li> <li>Prevent concentration in hollows and sumps.</li> <li>DO NOT enter confined spaces until atmosphere has been checked.</li> <li>Avoid smoking, naked lights, heat or ignition sources.</li> <li>When handling, DO NOT eat, drink or smoke.</li> <li>Vapour may ignite on pumping or pouring due to static electricity.</li> <li>DO NOT use plastic buckets.</li> <li>Earth and secure metal containers when dispensing or pouring product.</li> <li>Use spark-free tools when handling.</li> <li>Avoid contact with incompatible materials.</li> <li>Keep containers securely sealed.</li> <li>Avoid physical damage to containers.</li> <li>Always wash hands with soap and water after handling.</li> <li>Work clothes should be laundered separately.</li> <li>Use good occupational work practice.</li> <li>Observe manufacturer's storage and handling recommendations contained within this SDS.</li> <li>Atmosphere should be regularly checked against established exposure standards to ensure safe working conditions.</li> </ul>
Other information	<ul> <li>Store in original containers in approved flame-proof area.</li> <li>No smoking, naked lights, heat or ignition sources.</li> <li>DO NOT store in pits, depression, basement or areas where vapours may be trapped.</li> <li>Keep containers securely sealed.</li> <li>Store away from incompatible materials in a cool, dry well ventilated area.</li> <li>Protect containers against physical damage and check regularly for leaks.</li> <li>Observe manufacturer's storage and handling recommendations contained within this SDS.</li> </ul>

#### Conditions for safe storage, including any incompatibilities

Suitab	le container	<ul> <li>Packing as supplied by manufacturer.</li> <li>Plastic containers may only be used if approved for flammable liquid.</li> <li>Check that containers are clearly labelled and free from leaks.</li> <li>Manufactured product that requires stirring before use and having a viscosity of at least 20 cSt (25 deg. C): (i) Removable head packaging; (ii) Cans with friction closures and (iii) low pressure tubes and cartridges may be used.</li> </ul>
Storage inc	ompatibility	
+	×	



0 — May be stored together with specific preventions

+ — May be stored together

Note: Depending on other risk factors, compatibility assessment based on the table above may not be relevant to storage situations, particularly where large volumes of dangerous goods are stored and handled. Reference should be made to the Safety Data Sheets for each substance or article and risks assessed accordingly.

## **SECTION 8 Exposure controls / personal protection**

900 ppm

#### **Control parameters**

Occupational Exposure Limits (OEL)

## INGREDIENT DATA

Source	Ingredient	Material name	TWA	STEL		Peak	Notes	
Australia Exposure Standards	toluene	Toluene	50 ppm / 191 mg/m3	574 mg/m3 / 150	ppm	Not Available	Not Available	
Australia Exposure Standards	xylene	Xylene (o-, m-, p- isomers)	80 ppm / 350 mg/m3	655 mg/m3 / 150	ppm	Not Available	Not Available	
Emergency Limits								
Ingredient	TEEL-1		TEEL-2		TEEL-3	TEEL-3		
toluene	Not Available		Not Available	Not Available		Not Available		
xylene	Not Available		Not Available N		Not Ava	Not Available		
Ingredient	Original IDLH	Original IDLH		Revised IDLH				
toluene	500 ppm	500 ppm			Not Available			

Not Available

#### Exposure controls

xylene

	Engineering controls are used to remove a hazard or place be highly effective in protecting workers and will typically be The basic types of engineering controls are: Process controls which involve changing the way a job acti Enclosure and/or isolation of emission source which keeps 'adds' and 'removes' air in the work environment. Ventilatio ventilation system must match the particular process and c Employers may need to use multiple types of controls to pr For flammable liquids and flammable gases, local exhaust equipment should be explosion-resistant. Air contaminants generated in the workplace possess vary circulating air required to effectively remove the contaminant	e independent of worker interactio vity or process is done to reduce t a selected hazard 'physically' awa n can remove or dilute an air conta hemical or contaminant in use. event employee overexposure. ventilation or a process enclosure ing 'escape' velocities which, in tur	ns to provide this high level of protect he risk. y from the worker and ventilation th aminant if designed properly. The de ventilation system may be required.	ction. at strategically esign of a . Ventilation
	Type of Contaminant:			Air Speed:
	solvent, vapours, degreasing etc., evaporating from tank (	'in still air).		0.25-0.5 m/s (50-100 f/min.)
	aerosols, fumes from pouring operations, intermittent com plating acid fumes, pickling (released at low velocity into z		ransfers, welding, spray drift,	0.5-1 m/s (100-200 f/min.)
	direct spray, spray painting in shallow booths, drum filling, into zone of rapid air motion)	conveyer loading, crusher dusts,	gas discharge (active generation	1-2.5 m/s (200-500 f/min.)
Appropriate engineering controls	<ul> <li>Within each range the appropriate value depends on:</li> <li>Lower end of the range</li> <li>1: Room air currents minimal or favourable to capture</li> <li>2: Contaminants of low toxicity or of nuisance value only.</li> <li>3: Intermittent, low production.</li> <li>4: Large hood or large air mass in motion</li> <li>Simple theory shows that air velocity falls rapidly with dista with the square of distance from the extraction point (in sim accordingly, after reference to distance from the contamina 1-2 m/s (200-400 f/min.) for extraction of solvents generate considerations, producing performance deficits within the effactors of 10 or more when extraction systems are installed.</li> <li>Adequate ventilation is typically taken to be that which lim room or enclosure containing the dangerous substance.</li> <li>Ventilation for plant and machinery is normally considered potentially be present to no more than 25% of the LEL. Hos safeguards are provided to prevent the formation of a haza shutdown of the process might be used together with main turbine enclosures.</li> <li>Temporary exhaust ventilation systems may be provided to or other confined spaces or in an emergency after a releas atmosphere should be continuously monitored to ensure that space, the ventilation should ensure that the concentration provision of suitable breathing apparatus)</li> </ul>	aple cases). Therefore the air speet ting source. The air velocity at the d in a tank 2 meters distant from t xtraction apparatus, make it esser l or used. its the average concentration to n d adequate if it limits the average of wever, an increase up to a maximu rdous explosive atmosphere. For taining or increasing the exhaust v for non-routine higher-risk activitie e. The work procedures for such a at ventilation is adequate and the	d at the extraction point should be a extraction fan, for example, should he extraction point. Other mechanic natial that theoretical air velocities are o more than 25% of the LEL within t concentration of any dangerous subs im 50% LEL can be acceptable whe example, gas detectors linked to err entilation on solvent evaporating ov s, such as cleaning, repair or mainte ctivities should be carefully conside area remains safe. Where workers of the solution of the solution of the solution of the solution o	adjusted, be a minimum of al e multiplied by the building, stance that might ere additional hergency ens and gas enance in tanks red The will enter the



Eye and face protection	<ul> <li>Safety glasses with side shields.</li> <li>Chemical goggles.</li> <li>Contact lenses may pose a special hazard; soft contact lenses may absorb and concentrate irritants. A written policy document, describing the wearing of lenses or restrictions on use, should be created for each workplace or task. This should include a review of lens absorption and adsorption for the class of chemicals in use and an account of injury experience. Medical and first-aid personnel should be trained in their removal and suitable equipment should be readily available. In the event of chemical exposure, begin eye irrigation immediately and remove contact lens as soon as practicable. Lens should be removed at the first signs of eye redness or irritation - lens should be removed in a clean environment only after workers have washed hands thoroughly. [CDC NIOSH Current Intelligence Bulletin 59], [AS/NZS 1336 or national equivalent]</li> </ul>
Skin protection	See Hand protection below
Hands/feet protection	<ul> <li>Wear chemical protective gloves, e.g. PVC.</li> <li>Wear safety footwear or safety gumboots, e.g. Rubber</li> <li>NOTE:</li> <li>The material may produce skin sensitisation in predisposed individuals. Care must be taken, when removing gloves and other protective equipment, to avoid all possible skin contact.</li> <li>Contaminated leather items, such as shoes, belts and watch-bands should be removed and destroyed.</li> </ul>
Body protection	See Other protection below
Other protection	<ul> <li>Overalls.</li> <li>PVC Apron.</li> <li>PVC protective suit may be required if exposure severe.</li> <li>Eyewash unit.</li> <li>Ensure there is ready access to a safety shower.</li> <li>Some plastic personal protective equipment (PPE) (e.g. gloves, aprons, overshoes) are not recommended as they may produce static electricity.</li> <li>For large scale or continuous use wear tight-weave non-static clothing (no metallic fasteners, cuffs or pockets).</li> <li>Non sparking safety or conductive footwear should be considered. Conductive footwear describes a boot or shoe with a sole made from a conductive compound chemically bound to the bottom components, for permanent control to electrically ground the foot an shall dissipate static electricity from the body to reduce the possibility of ignition of volatile compounds. Electrical resistance must range between 0 to 500,000 ohms. Conductive should be stored in lockers close to the room in which they are worn. Personnel who have been issued conductive footwear should not wear them from their place of work to their homes and return.</li> </ul>

### Recommended material(s)

GLOVE SELECTION INDEX

Glove selection is based on a modified presentation of the:

'Forsberg Clothing Performance Index'.

The effect(s) of the following substance(s) are taken into account in the *computer-generated* selection: Altex Thinning Solvent #213

Altex Thinning Solve	nt #213	
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Material	CPI
BUTYL	С
BUTYL/NEOPRENE	С
CPE	С
HYPALON	С
NAT+NEOPR+NITRILE	С
NATURAL RUBBER	С
NATURAL+NEOPRENE	С
NEOPRENE	С
NEOPRENE/NATURAL	С
NITRILE	С
NITRILE+PVC	С
PE/EVAL/PE	С
PVA	С
PVC	С
PVDC/PE/PVDC	С
SARANEX-23	С
SARANEX-23 2-PLY	С
TEFLON	С
VITON	С
VITON/CHLOROBUTYL	С
VITON/NEOPRENE	С

\* CPI - Chemwatch Performance Index

A: Best Selection

B: Satisfactory; may degrade after 4 hours continuous immersion

C: Poor to Dangerous Choice for other than short term immersion

 $\ensuremath{\text{NOTE}}$  As a series of factors will influence the actual performance of the glove, a final selection must be based on detailed observation. -

\* Where the glove is to be used on a short term, casual or infrequent basis, factors such as 'feel' or convenience (e.g. disposability), may dictate a choice of gloves which might otherwise be unsuitable following long-term or frequent use. A qualified practitioner should be consulted.

Required Minimum Protection Factor	Half-Face Respirator	Full-Face Respirator	Powered Air Respirator
up to 10 x ES	A P1 Air-line*	-	A PAPR-P1 -
up to 50 x ES	Air-line**	A P2	A PAPR-P2
up to 100 x ES	-	A P3	-
		Air-line*	-
100+ x ES	-	Air-line**	A PAPR-P3

Type A-P Filter of sufficient capacity. (AS/NZS 1716 & 1715, EN 143:2000 & 149:2001,

\* - Negative pressure demand \*\* - Continuous flow

**Respiratory protection** 

ANSI Z88 or national equivalent)

A(All classes) = Organic vapours, B AUS or B1 = Acid gasses, B2 = Acid gas or hydrogen cyanide(HCN), B3 = Acid gas or hydrogen cyanide(HCN), E = Sulfur dioxide(SO2), G = Agricultural chemicals, K = Ammonia(NH3), Hg = Mercury, NO = Oxides of nitrogen, MB = Methyl bromide, AX = Low boiling point organic compounds(below 65 degC)

## **SECTION 9** Physical and chemical properties

Information on basic physical and chemical properties			
Appearance	paste		
Physical state	Free-flowing Paste	Relative density (Water = 1)	0.9
Odour	Not Available	Partition coefficient n-octanol / water	Not Available
Odour threshold	Not Available	Auto-ignition temperature (°C)	479
pH (as supplied)	Not Available	Decomposition temperature (°C)	Not Available
Melting point / freezing point (°C)	Not Available	Viscosity (cSt)	Not Available
Initial boiling point and boiling range (°C)	110	Molecular weight (g/mol)	Not Available
Flash point (°C)	4	Taste	Not Available
Evaporation rate	2 BuAC = 1	Explosive properties	Not Available
Flammability	HIGHLY FLAMMABLE.	Oxidising properties	Not Available
Upper Explosive Limit (%)	7.0	Surface Tension (dyn/cm or mN/m)	Not Available
Lower Explosive Limit (%)	1.2	Volatile Component (%vol)	92
Vapour pressure (kPa)	3	Gas group	Not Available
Solubility in water	Immiscible	pH as a solution (1%)	Not Available
Vapour density (Air = 1)	3.1	VOC g/L	828.18

# **SECTION 10 Stability and reactivity**

Reactivity	See section 7
Chemical stability	<ul> <li>Unstable in the presence of incompatible materials.</li> <li>Product is considered stable.</li> <li>Hazardous polymerisation will not occur.</li> </ul>
Possibility of hazardous reactions	See section 7
Conditions to avoid	See section 7
Incompatible materials	See section 7
Hazardous decomposition products	See section 5

## **SECTION 11 Toxicological information**

### Information on toxicological effects

Inhaled	The material can cause respiratory irritation in some persons. The body's response to such irritation can cause further lung damage. Inhalation of vapours may cause drowsiness and dizziness. This may be accompanied by sleepiness, reduced alertness, loss of reflexes, lack of co-ordination, and vertigo. Inhalation of high concentrations of gas/vapour causes lung irritation with coughing and nausea, central nervous depression with headache and dizziness, slowing of reflexes, fatigue and inco-ordination. Central nervous system (CNS) depression may include general discomfort, symptoms of giddiness, headache, dizziness, nausea, anaesthetic effects, slowed reaction time, slurred speech and may progress to unconsciousness. Serious poisonings may result in respiratory depression and may be fatal. Inhalation of vapours or aerosols (mists, fumes), generated by the material during the course of normal handling, may be damaging to the health of the individual.
Ingestion	At sufficiently high doses the material may be hepatotoxic (i.e. poisonous to the liver). Swallowing of the liquid may cause aspiration into the lungs with the risk of chemical pneumonitis; serious consequences may result. (ICSC13733) The material has <b>NOT</b> been classified by EC Directives or other classification systems as 'harmful by ingestion'. This is because of the lack of corroborating animal or human evidence. Considered an unlikely route of entry in commercial/industrial environments. The liquid may produce gastrointestinal discomfort and may be harmful if swallowed. Accidental ingestion of the material may be harmful; animal experiments indicate that ingestion of less than 150 gram may be fatal or may produce serious damage to the health of the individual.

#### Altex Thinning Solvent #213

Skin Contact	The material may accentuate any pre-existing dermatitis condition Skin contact with the material may damage the health of the individual; systemic effects may result following absorption. Open cuts, abraded or irritated skin should not be exposed to this material Entry into the blood-stream, through, for example, cuts, abrasions or lesions, may produce systemic injury with harmful effects. Examine the skin prior to the use of the material and ensure that any external damage is suitably protected. The material may cause moderate inflammation of the skin either following direct contact or after a delay of some time. Repeated exposure can cause contact dermatitis which is characterised by redness, swelling and blistering.		
Eye	The liquid produces a high level of eye discomfort and is capable of causing pain and severe conjunctivitis. Corneal injury may develop, with possible permanent impairment of vision, if not promptly and adequately treated. There is evidence that material may produce eye irritation in some persons and produce eye damage 24 hours or more after instillation. Severe inflammation may be expected with pain.		
Chronic	Long-term exposure to respiratory irritants may result in airways disease, involving difficulty breathing and related whole-body problems. Skin contact with the material is more likely to cause a sensitisation reaction in some persons compared to the general population. Toxic: danger of serious damage to health by prolonged exposure through inhalation, in contact with skin and if swallowed. This material can cause serious damage if one is exposed to it for long periods. It can be assumed that it contains a substance which can produce severe defects. Ample evidence exists that this material directly causes reduced fertility Substance accumulation, in the human body, may occur and may cause some concern following repeated or long-term occupational exposure. There has been some concern that this material can cause cancer or mutations but there is not enough data to make an assessment.		
Altex Thinning Solvent #213	TOXICITY Not Available	IRRITATION Not Available	
		Not Available	
	ΤΟΧΙΟΙΤΥ	IRRITATION	
	Dermal (rabbit) LD50: 12124 mg/kg <sup>[2]</sup>	Eye (rabbit): 2mg/24h - SEVERE	
	Inhalation(Rat) LC50: >13350 ppm4h <sup>[2]</sup>	Eye (rabbit):0.87 mg - mild	
	Oral (Rat) LD50; 636 mg/kg <sup>[2]</sup>	Eye (rabbit):100 mg/30sec - mild	
toluene	Eye: adverse effect observed (irritating) <sup>[1]</sup>		
	Skin (rabbit):20 mg/24h-moderate		
	Skin (rabbit):500 mg - moderate		
	Skin: adverse effect observed (irritating) <sup>[1]</sup>		
		Skin: no adverse effect observed (not irritating) <sup>[1]</sup>	
	TOXICITY	IRRITATION	
	Dermal (rabbit) LD50: >1700 mg/kg <sup>[2]</sup>	Eye (human): 200 ppm irritant	
	101		
	Inhalation(Rat) LC50: 5000 ppm4h <sup>[2]</sup>	Eye (rabbit): 5 mg/24h SEVERE	
xylene	Inhalation(Rat) LC50: 5000 ppm4h <sup>[2]</sup> Oral (Mouse) LD50; 2119 mg/kg <sup>[2]</sup>	Eye (rabbit): 87 mg mild	
xylene		Eye (rabbit): 87 mg mild Eye: adverse effect observed (irritating) <sup>[1]</sup>	
xylene		Eye (rabbit): 87 mg mild Eye: adverse effect observed (irritating) <sup>[1]</sup> Skin (rabbit):500 mg/24h moderate	
xylene		Eye (rabbit): 87 mg mild Eye: adverse effect observed (irritating) <sup>[1]</sup>	

Astima-like symptoms may continue for months of even years after exposure to the material ends. This may be due to a non-alteriate condition known as reactive airways dysfunction syndrome (RADS) which can occur after exposure to high levels of highly irritating compound. Main criteria for diagnosing RADS include the absence of previous airways disease in a non-atopic individual, with sudden onset of persistent asthma-like symptoms within minutes to hours of a documented exposure to the irritant. Other criteria for diagnosis of RADS include a reversible airflow pattern on lung function tests, moderate to severe bronchial hyperreactivity on methacholine challenge testing, and the lack of minimal lymphocytic inflammation, without eosinophilia. RADS (or asthma) following an irritating inhalation is an infrequent disorder with rates related to the concentration of and duration of exposure to the irritating substance. On the other hand, industrial bronchitis is a disorder that occurs as a result of exposure due to high concentrations of irritating substance (often particles) and is completely reversible after exposure ceases. The disorder is characterized by difficulty breathing, cough and mucus production. The following information refers to contact allergens as a group and may not be specific to this product.

Contact allergies quickly manifest themselves as contact eczema, more rarely as urticaria or Quincke's oedema. The pathogenesis of contact eczema involves a cell-mediated (T lymphocytes) immune reaction of the delayed type. Other allergic skin reactions, e.g. contact urticaria, involve antibody-mediated immune reactions. The significance of the contact allergen is not simply determined by its sensitisation potential: the distribution of the substance and the opportunities for contact with it are equally important. A weakly sensitising substance which is widely distributed can be a more important allergen than one with stronger sensitising potential with which few individuals come into contact. From a clinical point of view, substances are noteworthy if they produce an allergic test reaction in more than 1% of the persons tested. Data demonstrate that during inhalation exposure, aromatic hydrocarbons undergo substantial partitioning into adipose tissues. Following cessation of exposure, the level of aromatic hydrocarbons in body fats rapidly declines. Thus, the aromatic hydrocarbons are unlikely to bioaccumulate in the body. Selective partitioning of the aromatic hydrocarbons into the non-adipose tissues is unlikely. No data is available regarding distribution following dermal absorption. However, distribution following this route of exposure is likely to resemble the pattern occurring with inhalation exposure.

Aromatics hydrocarbons may undergo several different Phase I dealkylation, hydroxylation and oxidation reactions which may or may not be followed by Phase II conjugation to glycine, sulfation or glucuronidation. However, the major predominant biotransformation pathway is typical of that of the alkylbenzenes and consists of: (1) oxidation of one of the alkyl groups to an alcohol moiety; (2) oxidation of the hydroxyl group to a carboxylic acid; (3) the carboxylic acid is then conjugated with glycine to form a hippuric acid. The minor metabolites can be expected to consist of a complex mixture of isomeric triphenols, the sulfate and glucuronide conjugates of dimethylbenzyl alcohols, dimethylbenzoic acids and

		avdroeerbong, these substances are likely to be
	dimethylhippuric acids. Consistent with the low propensity for bioaccumulation of aromatic h significant inducers of their own metabolism. The predominant route of excretion of aromatic hydrocarbons following inhalation exposure parent compound, or urinary excretion of its metabolites. When oral administration occurs, t hydrocarbons, presumably due to the first pass effect in the liver. Under these circumstance route of excretion.	involves either exhalation of the unmetabolized there is little exhalation of unmetabolized these
XYLENE	Reproductive effector in rats The material may produce severe irritation to the eye causing pronounced inflammation. Re produce conjunctivitis. The substance is classified by IARC as Group 3: <b>NOT</b> classifiable as to its carcinogenicity to humans. Evidence of carcinogenicity may be inadequate or limited in animal testing.	epeated or prolonged exposure to irritants may
	For toluene: Acute toxicity: Humans exposed to high levels of toluene for short periods of time experience from headaches to intoxication, convulsions, narcosis (sleepiness) and death. When inhaled nervous system depression, and in large doses has a narcotic effect. 60mL has caused deat congestion and bleeding of the lungs and kidney injury were all found on autopsy. Exposure to inhalation at a concentration of 600 parts per million for 8 hours resulted in the (a feeling of well-being), dilated pupils, convulsions and nausea. Exposure to 1000-30000 narcosis and death. Toluene can also strip the skin of lipids, causing skin inflammation. Subchronic/chronic effects: Repeat doses of toluene cause adverse central nervous system and the kidney. Adverse effects occur from both swallowing and inhalation. In humans, a rej	d or swallowed, toluene can cause severe central ath. Death of heart muscle fibres, liver swelling, same and more serious symptoms including euphoria parts per million (1-3%) has been reported to cause n effects and can damage the upper airway, the liver
Altex Thinning Solvent #213 & TOLUENE	and the kicking. Adverse energy sector from both swallowing and initiation. In mains, a re- nervous system is 88 parts per million. In one case, toluene caused heart sensitization and of the cerebellum was noted. Workers chronically exposed to toluene fumes have reported rec Developmental/Reproductive toxicity: Exposure to high levels of toluene can result in advers have indicated that high levels of toluene can also adversely affect the developing offspring to toluene before birth, as a result of solvent abuse by the mother, variable growth, a small h deficits, minor facial and limb abnormalities, and developmental delay were seen. Absorption: Studies in humans and animals have shown that toluene is easily absorbed through the skin. Distribution: Animal studies show that toluene may be distributed in the body fat, bone marror with lower levels in the blood, kidney and liver. Toluene has generally been found to accumm. Metabolism: Inhaled or ingested toluene may be metabolized to benzyl alcohol, after which acid. Benzoic acid is sometimes conjugated with glycine to form hippuric acid or reacted wit O-cresol and p-cresol formed by ring hydroxylation are considered minor metabolites. Excretion: Toluene is mainly (60-70%) excreted through the urine as hippuric acid. Benzoyl unchanged toluene through exhaled air also accounts for 10-20%. Excretion of hippuric acid	death. In several cases of "glue sniffing", damage to duced white cell counts. se effects in the developing foetus. Several studies in laboratory animals. In children who were exposed head, central nervous system dysfunction, attention ough the lungs and gastrointestinal tract, with much ow, spinal nerves, spinal cord and brain white matter, ulate in fatty tissue, and in highly vascularised tissues it is further oxidized to benzaldehyde and benzoic th glucuronic acid to form benzoyl glucuronide. glucuronide accounts for 10-20% of excretion, and
-	nervous system is 88 parts per million. In one case, toluene caused heart sensitization and of the cerebellum was noted. Workers chronically exposed to toluene fumes have reported rec Developmental/Reproductive toxicity: Exposure to high levels of toluene can result in advers have indicated that high levels of toluene can also adversely affect the developing offspring to toluene before birth, as a result of solvent abuse by the mother, variable growth, a small H deficits, minor facial and limb abnormalities, and developmental delay were seen. Absorption: Studies in humans and animals have shown that toluene is easily absorbed through the skin. Distribution: Animal studies show that toluene may be distributed in the body fat, bone marror with lower levels in the blood, kidney and liver. Toluene has generally been found to accumu Metabolism: Inhaled or ingested toluene may be metabolized to benzyl alcohol, after which acid. Benzoic acid is sometimes conjugated with glycine to form hippuric acid or reacted wit 0-cresol and p-cresol formed by ring hydroxylation are considered minor metabolites. Excretion: Toluene is mainly (60-70%) excreted through the urine as hippuric acid. Benzoyl	death. In several cases of "glue sniffing", damage to duced white cell counts. se effects in the developing foetus. Several studies in laboratory animals. In children who were exposed head, central nervous system dysfunction, attention ough the lungs and gastrointestinal tract, with much ow, spinal nerves, spinal cord and brain white matter, ulate in fatty tissue, and in highly vascularised tissues it is further oxidized to benzaldehyde and benzoic th glucuronic acid to form benzoyl glucuronide. glucuronide accounts for 10-20% of excretion, and d is usually complete within 24 hours of exposure.
TOLUENE	nervous system is 88 parts per million. In one case, toluene caused heart sensitization and of the cerebellum was noted. Workers chronically exposed to toluene fumes have reported reconstruction before bether toxicity: Exposure to high levels of toluene can result in advers have indicated that high levels of toluene can also adversely affect the developing offspring to toluene before birth, as a result of solvent abuse by the mother, variable growth, a small he deficits, minor facial and limb abnormalities, and developmental delay were seen. Absorption: Studies in humans and animals have shown that toluene is easily absorbed through the skin. Distribution: Animal studies show that toluene may be distributed in the body fat, bone marriw with lower levels in the blood, kidney and liver. Toluene has generally been found to accumu. Metabolism: Inhaled or ingested toluene may be metabolized to benzyl alcohol, after which acid. Benzoic acid is sometimes conjugated with glycine to form hippuric acid. Benzoic scid is mainly (60-70%) excreted through the urine as hippuric acid. Benzoyl unchanged toluene through exhaled air also accounts for 10-20%. Excretion of hippuric acid. The material may cause skin irritation after prolonged or repeated exposure and may produ	death. In several cases of "glue sniffing", damage to duced white cell counts. se effects in the developing foetus. Several studies in laboratory animals. In children who were exposed head, central nervous system dysfunction, attention ough the lungs and gastrointestinal tract, with much ow, spinal nerves, spinal cord and brain white matter, ulate in fatty tissue, and in highly vascularised tissues it is further oxidized to benzaldehyde and benzoic th glucuronic acid to form benzoyl glucuronide. glucuronide accounts for 10-20% of excretion, and d is usually complete within 24 hours of exposure.
TOLUENE TOLUENE & XYLENE	nervous system is 88 parts per million. In one case, toluene caused heart sensitization and of the cerebellum was noted. Workers chronically exposed to toluene fumes have reported rec Developmental/Reproductive toxicity: Exposure to high levels of toluene can result in advers have indicated that high levels of toluene can also adversely affect the developing offspring to toluene before birth, as a result of solvent abuse by the mother, variable growth, a small H deficits, minor facial and limb abnormalities, and developmental delay were seen. Absorption: Studies in humans and animals have shown that toluene is easily absorbed through the skin. Distribution: Animal studies show that toluene may be distributed in the body fat, bone marror with lower levels in the blood, kidney and liver. Toluene has generally been found to accumm Metabolism: Inhaled or ingested toluene may be metabolized to benzyl alcohol, after which acid. Benzoic acid is sometimes conjugated with glycine to form hippuric acid or reacted wit O-cresol and p-cresol formed by ring hydroxylation are considered minor metabolites. Excretion: Toluene is mainly (60-70%) excreted through the urine as hippuric acid. Benzoyl unchanged toluene through exhaled air also accounts for 10-20%. Excretion of hippuric acid	death. In several cases of "glue sniffing", damage to duced white cell counts. se effects in the developing foetus. Several studies in laboratory animals. In children who were exposed head, central nervous system dysfunction, attention ough the lungs and gastrointestinal tract, with much ow, spinal nerves, spinal cord and brain white matter, ulate in fatty tissue, and in highly vascularised tissues it is further oxidized to benzaldehyde and benzoic th glucuronic acid to form benzoyl glucuronide. glucuronide accounts for 10-20% of excretion, and d is usually complete within 24 hours of exposure.
TOLUENE TOLUENE & XYLENE Acute Toxicity	nervous system is 88 parts per million. In one case, toluene caused heart sensitization and of the cerebellum was noted. Workers chronically exposed to toluene fumes have reported reconstruction before before betring to toluene thigh levels of toluene can result in adverse have indicated that high levels of toluene can also adversely affect the developing offspring to toluene before birth, as a result of solvent abuse by the mother, variable growth, a small he deficits, minor facial and limb abnormalities, and developmental delay were seen. Absorption: Studies in humans and animals have shown that toluene is easily absorbed through the skin. Distribution: Animal studies show that toluene may be distributed in the body fat, bone marrow with lower levels in the blood, kidney and liver. Toluene has generally been found to accumu. Metabolism: Inhaled or ingested toluene may be metabolized to benzyl alcohol, after which acid. Benzoic acid is sometimes conjugated with glycine to form hippuric acid. Benzoic acid is sometimes allow are considered minor metabolites. Excretion: Toluene is mainly (60-70%) excreted through the urine as hippuric acid. Benzoyl unchanged toluene through exhaled air also accounts for 10-20%. Excretion of hippuric acid	death. In several cases of "glue sniffing", damage to duced white cell counts. se effects in the developing foetus. Several studies in laboratory animals. In children who were exposed head, central nervous system dysfunction, attention ough the lungs and gastrointestinal tract, with much ow, spinal nerves, spinal cord and brain white matter, ulate in fatty tissue, and in highly vascularised tissues it is further oxidized to benzaldehyde and benzoic th glucuronic acid to form benzoyl glucuronide. glucuronide accounts for 10-20% of excretion, and d is usually complete within 24 hours of exposure. the on contact skin redness, swelling, the production of the second
TOLUENE TOLUENE & XYLENE Acute Toxicity Skin Irritation/Corrosion	nervous system is 88 parts per million. In one case, toluene caused heart sensitization and of the cerebellum was noted. Workers chronically exposed to toluene fumes have reported reconstruction before birth, as a result of solvent abuse by the mother, variable growth, a small be deficits, minor facial and limb abnormalities, and developmental delay were seen. Absorption: Studies in humans and animals have shown that toluene is easily absorbed through the skin. Distribution: Animal studies show that toluene may be distributed in the body fat, bone marrow with lower levels in the blood, kidney and liver. Toluene has generally been found to accumu. Metabolism: Inhaled or ingested toluene may be metabolized to benzyl alcohol, after which acid. Benzoic acid is sometimes conjugated with glycine to form inpuric acid. Benzoyl unchanged toluene through exhaled air also accounts for 10-20%. Excretion of hippuric acid. The material may cause skin irritation after prolonged or repeated exposure and may produvesicles, scaling and thickening of the skin.	Adeath. In several cases of "glue sniffing", damage to duced white cell counts. se effects in the developing foetus. Several studies in laboratory animals. In children who were exposed head, central nervous system dysfunction, attention ough the lungs and gastrointestinal tract, with much ow, spinal nerves, spinal cord and brain white matter, ulate in fatty tissue, and in highly vascularised tissues it is further oxidized to benzaldehyde and benzoic th glucuronic acid to form benzoyl glucuronide. glucuronide accounts for 10-20% of excretion, and d is usually complete within 24 hours of exposure. Ince on contact skin redness, swelling, the production of

Data available to make classification

Species

Not Available

Value

Not Available

Source

Not Available

### **SECTION 12 Ecological information**

Toxicity

 Endpoint
 Endpoint

 Not Available
 Endpoint

 Endpoint
 Te

 Endpoint
 Te

	Endpoint	Test Duration (hr)	Species	Value	Source
	EC50	48h	Crustacea	3.78mg/L	5
toluene	NOEC(ECx)	168h	Crustacea	0.74mg/L	5
	LC50	96h	Fish	5-35mg/l	4
	EC50	96h	Algae or other aquatic plants	>376.71mg/L	4

Test Duration (hr)

Not Available

	Endpoint	Test Duration (hr)	Species	Value	Source
	EC50	72h	Algae or other aquatic plants	4.6mg/l	2
xylene	EC50	48h	Crustacea	1.8mg/l	2
	NOEC(ECx)	73h	Algae or other aquatic plants	0.44mg/l	2
	LC50	96h	Fish	2.6mg/l	2
	-				

Ec

Legend:

Extracted from 1. IUCLID Toxicity Data 2. Europe ECHA Registered Substances - Ecotoxicological Information - Aquatic Toxicity 4. US EPA, Ecotox database - Aquatic Toxicity Data 5. ECETOC Aquatic Hazard Assessment Data 6. NITE (Japan) - Bioconcentration Data 7. METI (Japan) - Bioconcentration Data 8. Vendor Data

## Persistence and degradability

toluene LOW (Half-life = 28 days) LOW (Half-life = 4.33 days)	
xylene HIGH (Half-life = 360 days) LOW (Half-life = 1.83 days)	

### **Bioaccumulative potential**

Biodecultulative potential	
Ingredient	Bioaccumulation
toluene	LOW (BCF = 90)
xylene	MEDIUM (BCF = 740)
Mobility in soil	
-	Mach 2004
Ingredient	Mobility
toluene	LOW (KOC = 268)

# **SECTION 13 Disposal considerations**

Waste treatment methods	
Product / Packaging disposal	<ul> <li>Containers may still present a chemical hazard/ danger when empty.</li> <li>Return to supplier for reuse/ recycling if possible.</li> <li>Otherwise: <ul> <li>If container can not be cleaned sufficiently well to ensure that residuals do not remain or if the container cannot be used to store the same product, then puncture containers, to prevent re-use, and bury at an authorised landfill.</li> <li>Where possible retain label warnings and SDS and observe all notices pertaining to the product.</li> <li>DO NOT allow wash water from cleaning or process equipment to enter drains.</li> <li>It may be necessary to collect all wash water for treatment before disposal.</li> <li>In all cases disposal to sewer may be subject to local laws and regulations and these should be considered first.</li> <li>Where in doubt contact the responsible authority.</li> <li>Recycle wherever possible.</li> <li>Consult manufacturer for recycling options or consult local or regional waste management authority for disposal if no suitable treatment or disposal facility can be identified.</li> <li>Dispose of by: burial in a land-fill specifically licensed to accept chemical and / or pharmaceutical wastes or Incineration in a licensed apparatus (after admixture with suitable combustible material).</li> <li>Decontaminate empty containers. Observe all label safeguards until containers are cleaned and destroyed.</li> </ul> </li> </ul>

# **SECTION 14 Transport information**

#### Labels Required

Marine Pollutant	
HAZCHEM	•3YE

Land transport (ADG)

UN number	1263
UN proper shipping name	PAINT (including paint, lacquer, enamel, stain, shellac, varnish, polish, liquid filler and liquid lacquer base) or PAINT RELATED MATERIAL (including paint thinning or reducing compound)
Transport hazard class(es)	Class     3       Subrisk     Not Applicable
Packing group	II
Environmental hazard	Environmentally hazardous
Special precautions for user	Special provisions163 367Limited quantity5 L

## Air transport (ICAO-IATA / DGR)

UN number	1263
UN proper shipping name	Paint related material (including paint thinning or reducing compounds); Paint (including paint, lacquer, enamel, stain, shellac, varnish, polish, liquid filler and liquid lacquer base)

	ICAO/IATA Class	3		
Transport hazard class(es)	ICAO / IATA Subrisk	Not Applicable		
	ERG Code	3L		
Packing group	II			
Environmental hazard	Environmentally hazardous			
Special precautions for user	Special provisions Cargo Only Packing Instructions		A3 A72 A192 364	
	Cargo Only Maximum Qty / Pack		60 L	
	Passenger and Cargo Packing Instructions		353	
	Passenger and Cargo Maximum Qty / Pack		5 L	
	Passenger and Cargo Limited Quantity Packing Instructions		Y341	
	Passenger and Cargo Limited Maximum Qty / Pack		1 L	

### Sea transport (IMDG-Code / GGVSee)

UN number	1263		
UN proper shipping name	PAINT (including paint, lacquer, enamel, stain, shellac, varnish, polish, liquid filler and liquid lacquer base) or PAINT RELATED MATERIAL (including paint thinning or reducing compound)		
Transport hazard class(es)	IMDG Class     3       IMDG Subrisk     Not Applicable		
Packing group	I		
Environmental hazard	Marine Pollutant		
Special precautions for user	EMS NumberF-E, S-ESpecial provisions163 367Limited Quantities5 L		

## Transport in bulk according to $\ensuremath{\mathsf{Annex}}\xspace$ II of $\ensuremath{\mathsf{MARPOL}}\xspace$ and the IBC code

Not Applicable

## Transport in bulk in accordance with MARPOL Annex V and the IMSBC Code

Product name	Group	
toluene	Not Available	
xylene	Not Available	

#### Transport in bulk in accordance with the ICG Code

Product name	Ship Type
toluene	Not Available
xylene	Not Available

# **SECTION 15 Regulatory information**

## Safety, health and environmental regulations / legislation specific for the substance or mixture

## toluene is found on the following regulatory lists

Australia Hazardous Chemical Information System (HCIS) - Hazardous Chemicals	Australian Inventory of Industrial Chemicals (AIIC)
Australia Standard for the Uniform Scheduling of Medicines and Poisons (SUSMP) -	Chemical Footprint Project - Chemicals of High Concern List
Schedule 5	International Agency for Research on Cancer (IARC) - Agents Classified by the IARC
Australia Standard for the Uniform Scheduling of Medicines and Poisons (SUSMP) -	Monographs
Schedule 6	
xylene is found on the following regulatory lists	
Australia Hazardous Chemical Information System (HCIS) - Hazardous Chemicals	Australian Inventory of Industrial Chemicals (AIIC)
Australia Standard for the Uniform Scheduling of Medicines and Poisons (SUSMP) -	International Agency for Research on Cancer (IARC) - Agents Classified by the IARC
Schedule 5	Monographs
Australia Standard for the Uniform Scheduling of Medicines and Poisons (SUSMP) -	
Schedule 6	
National Inventory Status	

National Inventory	Status
Australia - AIIC / Australia Non-Industrial Use	Yes
Canada - DSL	Yes
Canada - NDSL	No (toluene; xylene)
China - IECSC	Yes

National Inventory	Status
Europe - EINEC / ELINCS / NLP	Yes
Japan - ENCS	Yes
Korea - KECI	Yes
New Zealand - NZIoC	Yes
Philippines - PICCS	Yes
USA - TSCA	Yes
Taiwan - TCSI	Yes
Mexico - INSQ	Yes
Vietnam - NCI	Yes
Russia - FBEPH	Yes
Legend:	Yes = All CAS declared ingredients are on the inventory No = One or more of the CAS listed ingredients are not on the inventory. These ingredients may be exempt or will require registration.

#### **SECTION 16 Other information**

Revision Date	10/11/2022
Initial Date	08/12/2017

#### SDS Version Summary

Version	Date of Update	Sections Updated
2.7	10/11/2022	Acute Health (eye), Acute Health (inhaled), Acute Health (skin), Acute Health (swallowed), Advice to Doctor, Chronic Health, Classification, Environmental, Exposure Standard, First Aid (inhaled), First Aid (skin), First Aid (swallowed), Ingredients, Personal Protection (Respirator), Physical Properties, Storage (storage incompatibility), Supplier Information, Use

#### Other information

Classification of the preparation and its individual components has drawn on official and authoritative sources as well as independent review by the Chemwatch Classification committee using available literature references.

The SDS is a Hazard Communication tool and should be used to assist in the Risk Assessment. Many factors determine whether the reported Hazards are Risks in the workplace or other settings. Risks may be determined by reference to Exposures Scenarios. Scale of use, frequency of use and current or available engineering controls must be considered.

#### Definitions and abbreviations

PC-TWA: Permissible Concentration-Time Weighted Average PC-STEL: Permissible Concentration-Short Term Exposure Limit IARC: International Agency for Research on Cancer ACGIH: American Conference of Governmental Industrial Hygienists STEL: Short Term Exposure Limit TEEL: Temporary Emergency Exposure Limit. IDLH: Immediately Dangerous to Life or Health Concentrations ES: Exposure Standard OSF: Odour Safety Factor NOAEL :No Observed Adverse Effect Level LOAEL: Lowest Observed Adverse Effect Level TLV: Threshold Limit Value LOD: Limit Of Detection OTV: Odour Threshold Value BCF: BioConcentration Factors BEI: Biological Exposure Index AIIC: Australian Inventory of Industrial Chemicals DSL: Domestic Substances List NDSL: Non-Domestic Substances List IECSC: Inventory of Existing Chemical Substance in China EINECS: European INventory of Existing Commercial chemical Substances ELINCS: European List of Notified Chemical Substances NLP: No-Longer Polymers ENCS: Existing and New Chemical Substances Inventory KECI: Korea Existing Chemicals Inventory NZIoC: New Zealand Inventory of Chemicals PICCS: Philippine Inventory of Chemicals and Chemical Substances TSCA: Toxic Substances Control Act TCSI: Taiwan Chemical Substance Inventory INSQ: Inventario Nacional de Sustancias Químicas NCI: National Chemical Inventory FBEPH: Russian Register of Potentially Hazardous Chemical and Biological Substances

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